AQUIFER STUDIES RELATED TO GEOLOGIC STORAGE OF CO₂

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CCS FACTS

Carbon capture and storage (CCS) is a process that will be used to reduce atmospheric emissions from coal-fired power plants across the globe
Geologic sequestration (GS) is a method of isolating CO₂ in the deep subsurface via injection of compressed fluid and

subsequent trapping

 A properly selected GS site should not impact underground sources of drinking water (USDWs), <10,000 mg/L TDS

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Monitoring is necessary!



CONSIDERATIONS

- Types of monitoring Really want to look for early warnings!
 - Deep subsurface look for pressure changes just above low permeability seal overlying injection zone
 - Intermediate depth subsurface base of USDWs and below
 - Shallow subsurface saturated and unsaturated zones
 - Surface/Atmospheric a way to confirm sequestration credits
- Displacement/migration of high salinity fluids (brine) with dissolved CO₂ by conduit flow along high permeability pathways:
 - insufficiently plugged well bores
 - transmissive faults
 - upward along gently dipping, regional water-bearing strata (i.e. Carrizo-Wilcox aquifer)
- Different aquifers contain different matrix minerals that will react with CO₂ at different rates dissolution and precipitation
- Shallow groundwater contains natural sources of CO₂ NOT A HAZARDOUS WASTE!

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WATER QUALITY ISSUES

CO₂ injected into a water-bearing formation (freshwater or brine) will partially dissolve and:

Reduce pH of solution

 $\begin{array}{l} \mathsf{CO}_{2(g)} + \mathsf{H}_2\mathsf{O} \leftrightarrow \mathsf{H}_2\mathsf{CO}_{3(aq)} \\ \mathsf{H}_2\mathsf{CO}_{3(aq)} \leftrightarrow \mathsf{HCO}_{3(aq)}^- + \mathsf{H}^+ \\ \mathsf{HCO}_{3(aq)}^- \leftrightarrow \mathsf{CO}_{3(aq)}^{2-} + \mathsf{H}^+_{(aq)} \end{array}$

Chemically interact with formation matrix to:

- increase dissolved cation concentrations (Group 1 - B, Ba, Ca, Co, K, Mg, Mn, Sr, and Zn)
- increase dissolved cations followed by concentration decline (Group 2 – Al, As, Cs, Cu, Fe, Mo, Ni, Rb, U, and V).

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CASE STUDY – SACROC Oilfield



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CASE STUDY – Dockum Aquifer



Not all of these wells are completed in the Dockum aquifer!

Six groundwater monitoring trips – June 2006, July 2007, November 2007, March 2008, July 2008, November 2008, and hopefully July 2009

BEG sampled 60 private water wells, 1 freshwater spring, and brine from 8 CO₂ injection zone wells – total 123 samples

TWDB database water quality data for analyses with charge balance error <10%, potassium analysis, and good reliability code

1909-2009

CASE STUDY - Challenges



Need Landowner Permission for Access to Private Water Wells



BEG and TWDB Water Well Data

Geologic units

Q – Quaternary undifferentiated

P-Eog – Paleocene-Eocene Ogallala

TrD – Triassic Dockum

P – Permian undifferentiated

TWDB wells **BEG** wells SACROC oilfield Ρ TrD Drainage Basin Scurry County Divide Colorado River TrD Miles

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Surface geology from BEG Big Spring and Lubbock GAT sheets

Stratigraphic Cross-sections from Shallow NW Geophysical Logs SE



NW-SE Cross Section Based on TCEQ Surface Casing Geophysical Logs





Cross section generated in GeoGraphix and Adobe Illustrator software using scanned logs from TCEQ Surface Casing Division log library.

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Potentiometric Surface Map for Scurry and nearby counties – only wells with total depth (TD) in Dockum Santa Rosa



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pH Along Gradientparallel Transects

All BEG-sampled wells completed through Ogallala with TD in Dockum Santa Rosa



909-2009





Line A-A' – Example Well Transect for Reactive Flow Path Modeling





Line A-A' – Example Well Transect for Reactive Flow Path Modeling











SACROC AREA WATER QUALITY

60 wells and 1 spring; filtered cations; unfiltered anions; highest concentration measured in each well

	EPA/TCEQ		BEG Wells	BEG Wells
	Primary Drinking	BEG Wells	Exceeding EPA	Exceeding EPA
	Water	Exceeding EPA	Standards -	Standards -
Analyte	MCL (mg/L)	Standards	Inside SACROC	Outside SACROC
Brimany Mavimum Contaminant Loval (MCL)				
Arsenic (As)	0.01	9.8 %	1.6 %	8.2 %
Cadmium (Cd)	0.005	1.6 %	0.0 %	1.6 %
Fluoride (F ⁻)	0.4	4.9 %	1.6 %	3.3 %
Nitrate (NO ₃ -N)	10	13.1 %	4.9 %	8.2 %
Selenium (Se)	0.05	4.9 %	1.6 %	3.3 %
Secondary Drinking Water Standard				
Alumnium (Al)	0.05	34.4 %	13.1 %	21.3 %
Chloride (Cl ⁻)	250	32.8 %	14.8 %	18.0 %
Fluoride (F ⁻)	0.2	37.7 %	14.7 %	23.0 %
Manganese (Mn)	0.05	14.7 %	4.9 %	9.8 %
Sulfate (SO ₄ ²⁻)	250	26.2 %	1.6 %	24.6 %
Total Dissolved Solids (TDS)	1000	50.8 %	18.0 %	32.8 %

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CONCLUSIONS



• No distinction between Dockum water quality inside and outside of SACROC

 Dockum is a heterogeneous groundwater system

• Not clear if we can – or will be able to - see evidence of injectate CO₂ in Dockum groundwater

• Good news for GS of CO₂ at least at SACROC

• Good job Kinder Morgan!

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BEG Water Well Monitoring

- Six sampling trips Fisher Co. in June 2006; Scurry, Garza, and Kent counties in July 2007, November 2007, March 2008, July 2008, and November 2008
- Total wells 60 wells and 1 spring (+8 SACROC production and brine injection wells)
- Total samples sets collected 123
- Laboratory analytes (LANL): Al, Ag, As, B, Ba, Be, Br, Ca, Cd, Cl, Co, CO₃, Cr, Cs, Cu, d¹³C, dD, d¹⁸O, F, Fe, HCO₃, Hg, K, Li, Mg, Mn, Mo, Na, Ni, NO₃, Pb, PO₄, Rb, Sb, Se, Si, Sn, SO₄, Sr, TDS, Th, Ti, Tl, U, V, and Zn
- Laboratory analytes (UT DGS Romanak): DIC, DOC, headspace gases (CO₂, CH₄)
- Field parameters: dissolved oxygen, pH, specific conductivity, and temperature; alkalinity titrations
- Well information: total depth, water level (where possible), x and y coordinates from GPS, elevation (z) from digital elevation model, stratigraphic unit from BEG-constructed structure contour maps (based on shallow geophysical logs)

