# CCS Safety and Environmental Impacts

#### **Global Efforts**

Since 2006, there have been many field experiments across the globe simulating a CO<sub>2</sub> leak in order to understand potential environmental impacts.

Click/Scan QR code for more info.

25 years of lab and field research and deployment show that CCS sites are unlikely to leak, but if they did, we have to understand and consider potential risks. Here is an overview.



## **Human Health**

Potential  $CO_2$  impacts include contamination of drinking water, displacement of oxygen in low-lying areas, and threats to ecosystem health. While  $CO_2$  is not explosive or toxic, it can displace oxygen, posing a risk to health via oxygen deprivation. However, the overall risk is lower than everyday risks such as driving a car or being struck by lightning.



# **Potential Groundwater Impacts**

 $\rm CO_2$  can potentially impact groundwater by causing the mobilization of heavy metals due to mineral dissolution or detachment from the grain surface, leading to contamination. Additionally, brine, which is difficult to clean up, can render water undrinkable due to its salt content.



# **Metal Mobilization**

Research in both lab and in field experiments show that while  $CO_2$  can mobilize metals in groundwater, the effect is often minimal and transient. The mobilized metals usually fall back out once the  $CO_2$  or the groundwater moves away, and the amount is typically not enough to affect drinking water standards. Therefore, metal mobilization is no longer a major concern.



# **Brine Migration Potential**

The main concern is brine contamination in groundwater, which is difficult to remedy. This risk is associated with closure and boundary conditions, and abandoned wells. It's crucial to manage injection pressure and ensure abandoned wells are properly plugged to prevent brine from being pushed into the aquifer.



#### **Terrestrial Ecosystem**

Ecosystem impacts of  $CO_2$  are spatially limited and ecosystems have existing uptake mechanisms for  $CO_2$ , as it is not a toxic contaminant but a natural ecosystem component has been encountered before. Certain plants can tolerate high levels of  $CO_2$ . Over time there may be a shift to more acid-tolerant species.



# Marine Ecosystem

Marine Ecosystems exhibit resilience to  $CO_2$  leaks, with most organisms able to tolerate or escape the effects. The main damage is to organisms with calcium shells, particularly in their larval stage. However, the ecosystem's familiarity with  $CO_2$  reduces the overall impact.

TEXAS Geosciences The University of Texas at Austin Jackson School of Geosciences Bureau of Economic Geology



### Did you Know?

There have been no instances of carbon dioxide leakage from a deep storage formation to the groundwater or to the surface.

Controlled release experiments around the world have shown that  $CO_2$  leakage would not be as disastrous as once thought, with impacts being short-lived and spatially small.

Controlled Release Projects

Read: What have we learnt about CO<sub>2</sub> leakage from CO<sub>2</sub> release field experiments, and what are the gaps for the future?



J. Roberts and L. Stalker (2020)

Secure storage - why CO2 doesn't leak

 $CO_2$  is stored in liquid-like form at least 800 meters (2600 ft) deep. It is trapped in the pores of the rock like a stain is trapped in your clothing.

The main avenue for leakage is wells, often referred to as artificial penetrations, which provide a path for migration.

Source: Dr. Katherine Romanak

